





variety of orange reproduce in very large amount in the Kazan area in Southern Turk (14). Grapefruit wine is differ in flavor when compared with orange wine because of the metabolism of the fruits volatile and secretion of yeast volatile during fermentation (15).

Flavoring compounds are the key contributor in determining the final product of fermentation as they having much medicinal value (16). Phytochemicals are present in fruit plants and any other part of the plant, which can be considered nutritive and non- nutritive biological compound. Phytochemicals also having health-related beneficial effects (17,18). Various types of Phytochemicals such as Anthraquinone, flavonoid, protein, carbohydrate, phenolic compound etc. Phytoalexins is one of the unique and delicious class of phytochemicals. Its uniqueness is, its good response towards external stress i.e. draught, sunburn, etc. A stipend type of constituent is having effectiveness against heart diseases that is a cardiovascular disease also having a positive impact against the tumor, and they are generally used in juices of red grapes wine (19). Physiochemical tests are commonly used to determine wine certification. These test include determination of specific gravity, alcohol content, etc.(20,21,22)

## **2. MATERIAL AND METHOD :**

### **2.1 Procurement of raw material and must Fermentation:**

Three different fresh fruits were used which include Carrot (*Daucuscarrota*), Papaya (*Carica papaya*), and Orange (*Citrus sinensis*) for making wine, which has been purchased from the local market of Panchkula.

#### **2.1.1. The exaction of fruit juice (Fermentation media)**

The fruits were washed thoroughly with sterile water and rinsed to remove unwanted material. Pulp from papaya was extracted and filtered with a muslin cloth. Fresh juice of orange and carrot was also extracted by filtration.

#### **2.1.2. Must Preparation:**

Fruit juices from papaya, orange, and carrot has been taken and mixed in 10:17:9 ratios respectively. The fruits were cut, manually deseeded, blended and filtered to obtain the must. The must was then transferred to a sterile 500 ml flask, yeast has been inoculated ( as a starter culture) and incubated at 42°C for 24 days with the analysis of parameters such as pH, temperature, Specific gravity, Titrable acidity, sugar content and ethanol content.

## 2.2 PHYSIOCHEMICAL ANALYSIS OF WINE

Physiochemical parameters were routinely analyzed at a regular interval of 3 days. These parameters included the monitoring of pH, sugar, temperature, specific gravity, Titrable acidity and ethanol content. These tests usually help in fixing the wine quality (for wine consumer as per health concern).

### 2.2.1 pH determination

The pH scale is used to determine the acidic and basic strength of any solution. pH meter has been used to check the pH. During wine fermentation decrease in pH has been observed because of the consumption of sugar in used fruits.

### 2.2.2 Sugar Content

The content of sugar was measured with the help of refractometer routinely. Change in sugar content was recorded Sugar is utilized by the yeast for the production of ethanol.

### 2.2.3 Specific gravity

It is measure as the ratio of the density of a substance to the density of a reference substance. For which special specific gravity bottle has been used. 25 ml specific gravity bottle was thoroughly cleaned with distilled water then put in a hot air oven at 50°C for drying. After that weight of the bottle was recorded as ( $W_1$ ) and filled it with distilled water. Surface of the bottle was cleaned with a wall of cotton wool and weighed as ( $W_2$ ). Thereafter the bottle was filled with the “must” and weighed as ( $W_3$ ). With the help of recorded values, the specific gravity was calculated (23).

$$S. G = \frac{W_3 - W_1}{W_2 - W_1} = \underline{S}$$

$$W_2 - W_1 = W$$

Where  $S$  = Weight of the sample ( $W_3 - W_1$ )

$$W = \text{Weight of the volume of water } (W_2 - W_1)$$

### 2.2.4 Titrable Acidity

Titrable acidity is a measure of the amount of acid present in the given sample. It was estimated by titration method by the use of titrant i.e., Sodium hydroxide (0.1N) which is titrated against sample (wine), in a conical flask 5ml of wine sample and 20 ml of distilled water along with 2 to 3 drops of Phenolphthalein indicator. Phenolphthalein was used as an indicator. (24).

Which is calculated as:

$$T.A. = \frac{\text{amount of NaOH used}}{\text{the Volume of sample taken}} \times 0.75$$

the Volume of sample taken

### ***2.2.5 Ethanol content***

The quantity of ethanol content was estimated by using hydrometer. Generally, ethanol content was increased during fermentation process because of conversion of sugar into ethanol due to yeast activity.

## **2.3 PHYTOCHEMICAL ANALYSIS**

Phytochemicals are non-nutritive plant chemicals that usually have protective or disease preventive properties such chemicals show a rare side-effect on a living being. Many organic compounds present in different parts of plants like in roots, shoots, fruits etc. they impart positive physiological impact on human beings, flavonoid, alkaloid, tannin and many more are such bioactive compound. (25,26). A large group of Phytochemicals have been shown to have an inhibitory effect on all type of microorganism. (27).

### ***2.3.1 Test for Anthraquinone***

1 ml of wine sample was added in 1 ml of conc. Sulphuric acid ( $H_2SO_4$ ) and 1 ml benzene in a test tube. Then 1 ml diluted Ammonia was added. Appearance of rose-pink color will indicate the positive result (28).

### ***2.3.2 Test for flavonoid***

#### ***Shinoda test***

The wine sample was added with few fragments of magnesium ribbon and drop wise concentrated HCL was added if after few minutes pink color scarlet observed. This was taken as the flavonoid presence (25).

### ***2.3.3 Test for protein***

**Ninhydrin's test:** In test tube wine was mixed with 02ml of 0.2% chemical solution of Ninhydrin, presence of amino acid and protein is confirmed if violet color appeared. (25).

### ***2.3.4 Test for carbohydrate***

#### ***Molisch Test***

Wine sample was mixed with 2ml of Molisch's reagent and the mixture was shaken properly. Then 2ml of concentrated Sulphuric acid was poured along the side of the test tube. Appearance of violet color shows the presence of carbohydrate (29).

### ***2.3.5 Test for tannin and phenol***

Wine sample was mixed with 2ml of 0.2% FeCl<sub>3</sub> solution and the appearance of blue-green or black coloration shows that both the organic component are present. (30).

### ***2.3.6 Test for alkaloid***

The wine was added with 2ml of 1% solution of HCL and was gently heated. Then after that Mayer's and Wanger's reagent were added to the mixture. Turbidity of the precipitate was taken as proof for the presence of alkaloid. (25).

### ***2.3.7 Test for saponins***

In test tube, few ml of wine was added and mixed with 5ml of distilled water and the mixture was shaken vigorously. The appearance of stable foam was considered as the indication towards saponins presence in the given sample. (31).

### ***2.3.8 Test for Terpenoid***

Wine was dissolved in 2ml of chloroform and evaporated to dryness than 2ml of concentrated Sulphuric acid was added and heated for about 2 minutes. A grayish color indicates the presence of terpenoid (32).

## **3. RESULT AND DISCUSSION**

### ***3.1 Physicochemical parameter analysis***

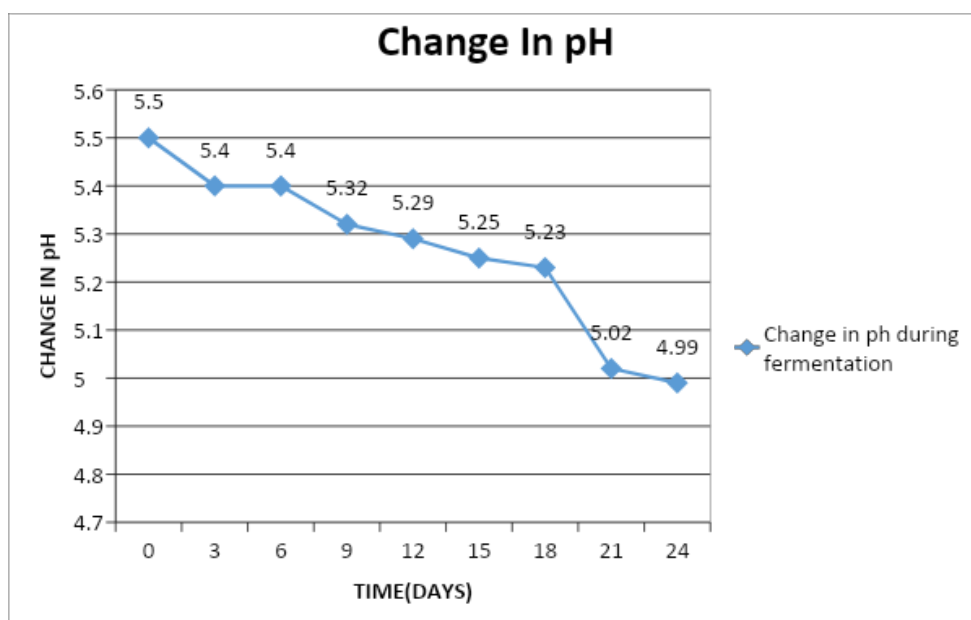
The following table is revealing the result of various parameter analysis determined during the fermentation of "must" after pitching with the yeast culture.

DAYS	Ph	Temp.	S.G.	T.A.	S.C.	E.C.
0	5.5	42	1.0783	0.002	20.5	0
3	5.4	29.9	1.0673	0.36	10	0.361
6	5.4	29	1.031	0.225	8	1.256
9	5.3	27	1.03	0.45	7.5	1.902
12	5.29	28	1.001	0.45	7	2.47
15	5.25	30	1.000	0.46	6.8	3.4
18	5.23	29	0.999	0.49	6	5.33
21	5.02	31	0.962	0.51	5.51	5.4
24	4.99	27	0.944	0.69	5	5.49

Table 1: Represents the record of parameters where Temperature, S.C.,T.A., S.G. and E.C. are temperature , Sugar content ,Titrable acidity, specific gravity and ethanol content respectively.

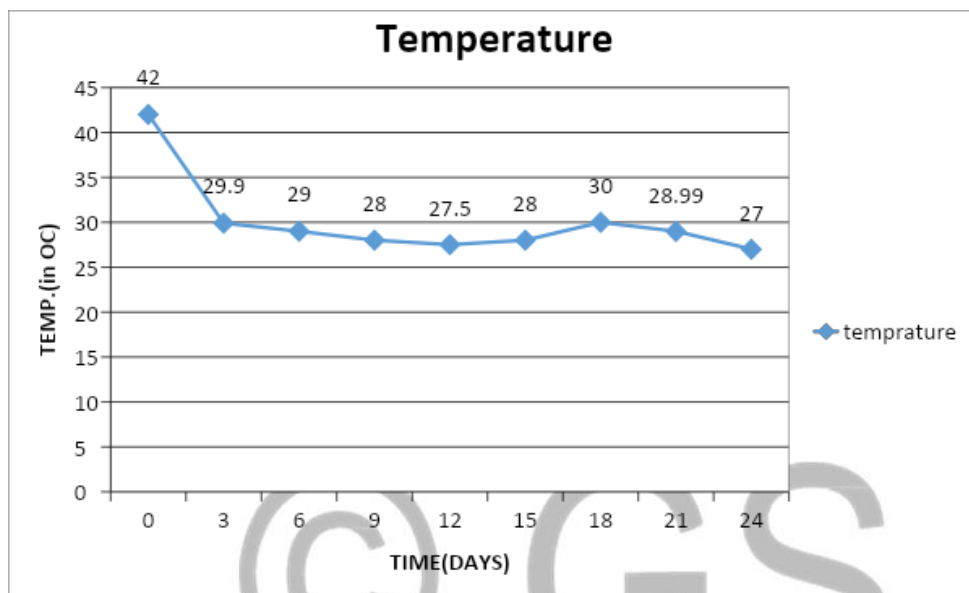
During the period of fermentation, various parameters like pH, temperature, specific gravity , Titrable acidity etc. were observed at a determined interval of time. There were fluctuation in these parameters were noted because of many factors like type of strain used and fermentation conditions during that time. Changes in these parameters are mention in further given graphical presentation.

**FIG.3.1 Change in pH**



The given graph in figure 4.1 shows a gradual decrease in pH during fermentation of wine, overall we observed from the initial day to 24<sup>th</sup> day of fermentation, the consumption of sugar by yeast activity and continue falling of pH from 5.5 to 4.99. And we saw a steep decrease from 5.23 to 5.02 on 21 day of fermentation, final day of the process, the observed pH was 4.99. This given graph is showing progressive reduction of pH.

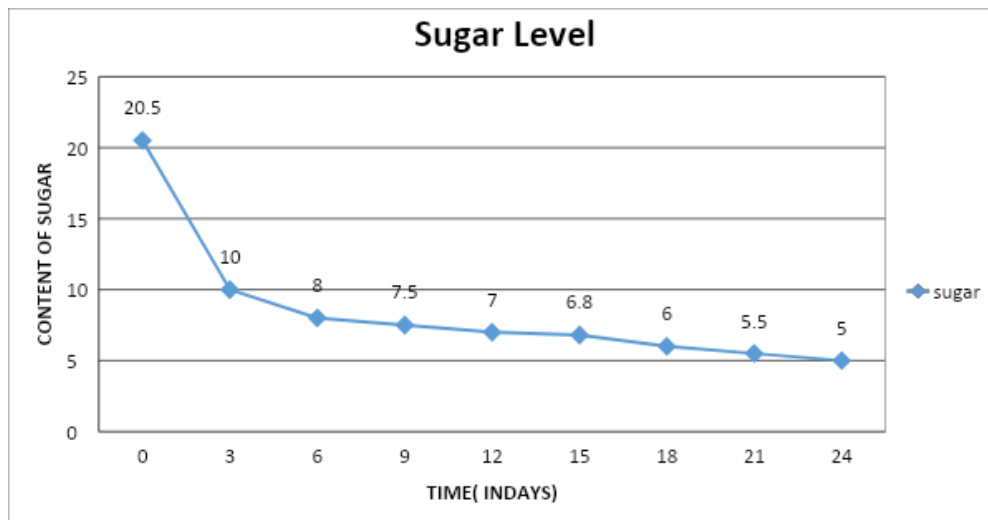
**FIG.3.2 Variation in temperature during fermentation**



The given graph indicates the change in temperature during must fermentation activities carried out by *Saccharomyces cerevisiae*. On the very first day we have observed the temperature was 42 degrees Celsius and then frequent decrease on the third day of fermentation was recorded. Slight fluctuation in temperature was observed between the 15<sup>th</sup> to 18<sup>th</sup> day and, on the final day its reached to 27<sup>o</sup>C. This fluctuation in temperature may was observed due to microbial activities.

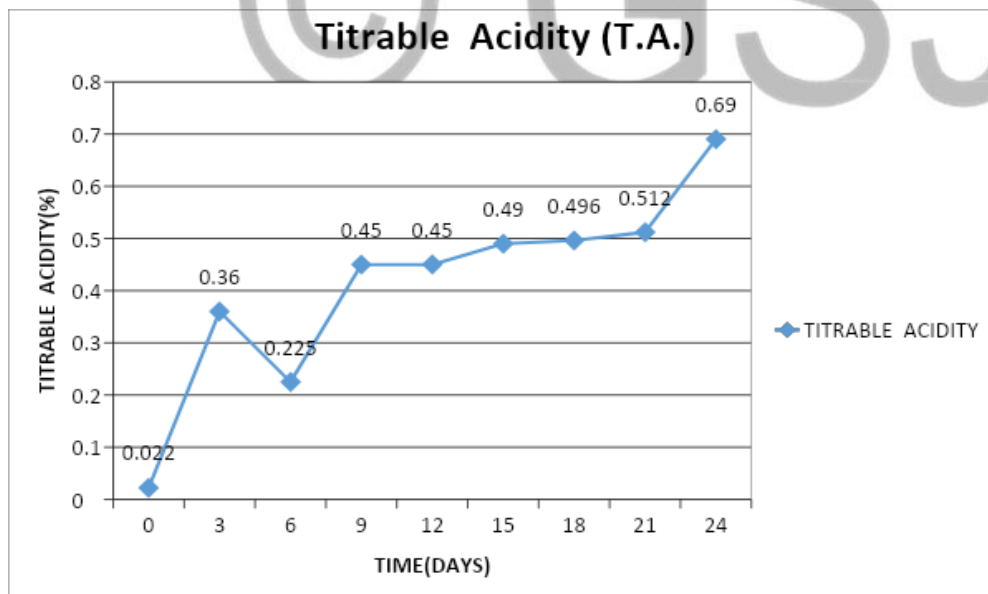


**FIG.3.3** Given graph is Showing change in sugar content during fermentation of wine from mixed fruits juice.



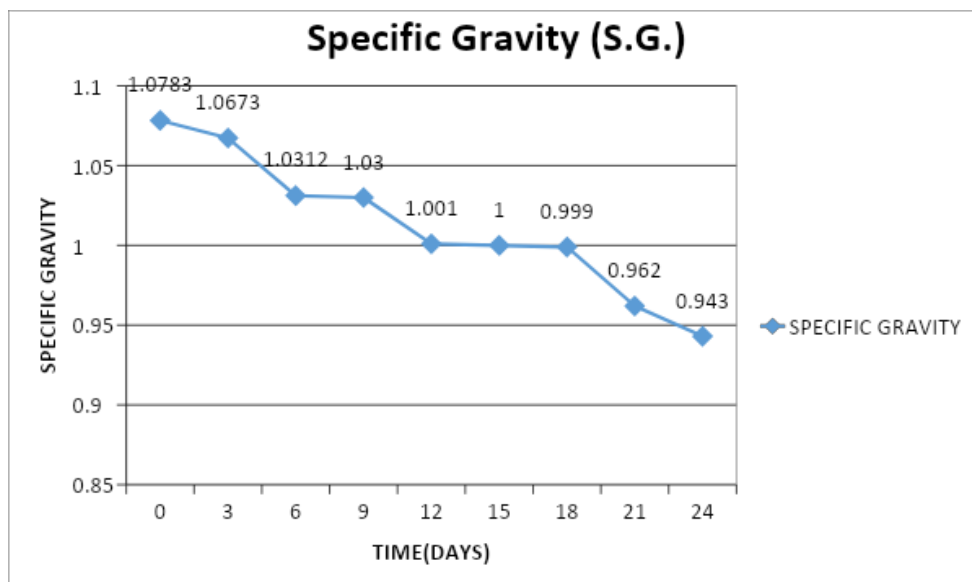
The given graph tells, decrease in the content of sugar of wine from initial stage to final stage. As we have observed reduction in sugar with the passage of fermentation day because of the utilization of sugar by the strain of yeast.

**FIG. 3.4** Change in Titrable Acidity during period of fermentation.



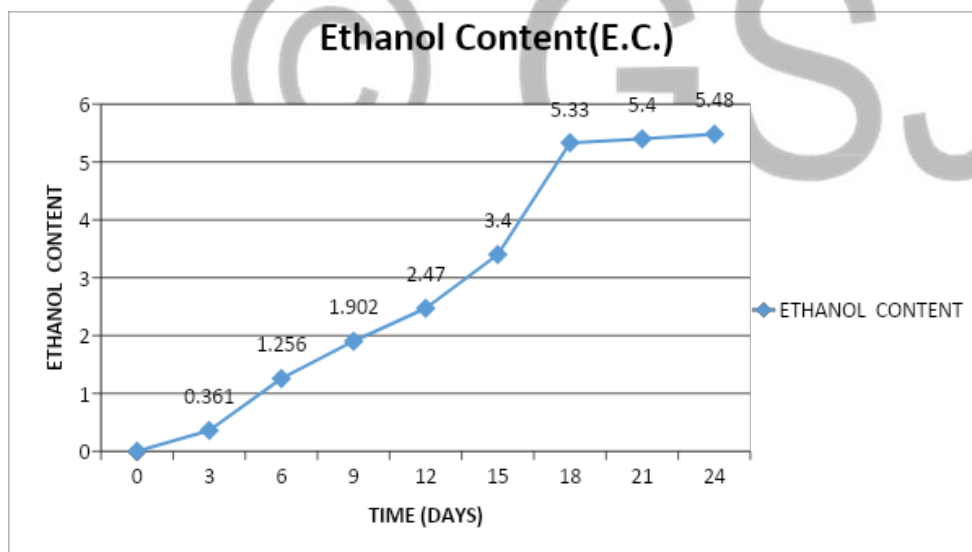
The given graph of titrable acidity shows fluctuation in the value of acidity during the 24 day of fermentation. On the first day acidity was 0.022 and on the last day of this process it reached at the value 0.69.

**FIG.3.5** Variation in Specific gravity during the process of fermentation.



This gives the information about the specific gravity observed during wine production. In mixed fruit wine gradual decrease in specific gravity was noticed from 1.0783 -0.944 in 24<sup>th</sup> day respectively.

**FIG. 3.6 Record of ethanol content during the period of wine fermentation.**



The given graph information illustrate that the percentage of ethanol is increased gradually with the passage of fermentation process. And on the final day of this process of conversion of sugar to Alcohol and carbon-dioxide by the action of yeast was 5.49%,so we have observed this percentage as the final yield of ethanol content.

The total content of alcohol was increased from initial day to the end of fermentation; this was 0.36% on day one and 5.55% on last day of the fermentation process. This information resembles

with the result of Okunowo, W.O., Okotore,R.O., & Osuntoki,A.A,(2005). The change in the` level of alcohol would be due the change of their Phytochemicals conditions and others factors such as temperature or pH (33,34). It also has been disclosed that the expression of gene-level in yeast is effected by environmental factors such as temperature (35).

## CONCLUSION:

Mixed fruit wines are undistilled alcoholic beverages made from basically grapes and other fruits like orange, papaya, carrot etc. Most of the fermented wines have nutritional values and their adequate amount of intake helps in reducing various disease related problems. The best example of such type of wine is red wine. Except for these benefits fermented wine also contain Phytochemicals as well as antioxidant. In this study the result are representing the decrease in content of sugar and simultaneously increase in ethanol content during the period of fermentation. Analysis of Phytochemicals disclosed the presence of alkaloid, flavonoid and terpenoid. The antioxidant and anticancerous study can be pursued and further investigation can be done with this project.

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