





### **Total Soluble Solid (TSS)**

This was determined using a hand refractometer. The TSS value was read directly from the calibrated scale of the refractometer. It was expressed as (%) sucrose or degree brix (El-Faki and Eisa, 2010).

### **Specific Gravity**

This was obtained using an hydrometer. It is expressed mathematically as

$$\frac{\rho_{wort}}{\rho_{water}}$$

Where  $\rho_{wort}$

is the density of the wort and  $\rho_{water}$  is the density of water (Fellows, 2005)

### **Determination of Nutritional Content of Natural Fermented *Obiolor***

#### **Moisture Determination**

Ten ml of sample was measured in a clean crucible using sensitive balance (Ekanem *et al.*, 2018). The samples were put into a moisture extraction oven at 105°C and heated for 3 h. The dried samples were put into desiccators, allowed to cool and reweighed. The procedure was repeated until steady weight was obtained.

#### **Ash Determination**

The ash content was determined from the loss in weight that occurred during incineration of the evaporated sample at a temperature high enough to permit all organic matter to be burnt off without allowing appreciable breakdown of the ash constituent. Ashing was carried out in a muffle furnace subjected to heat at 550°C for 6 h (A. O. A. C., 2005).

#### **Crude fibre Determination**

Two grams of the sample was transferred into a 1L conical flask. One hundred millilitres of sulfuric acid (0.255 mol/L) was heated to boiling and then introduced into the conical flask containing the sample. The contents were then boiled for 30 mins, ensuring that the level of the acid was maintained by the addition of distilled water. After 30 mins, the contents were then filtered through a muslin cloth held in a funnel. The residue was rinsed thoroughly until its washing was no longer acidic to litmus. The residue was then transferred into a conical flask. One hundred milliliters of sodium hydroxide (0.313 mol/L) was then brought to boil and then introduced into the conical flask containing the sample. The contents were then boiled for 30 mins, ensuring that the level of the acid was maintained by the addition of distilled water. After 30 mins, the contents were then filtered through a Whatman 125 mm filter paper held in a funnel. The residue was rinsed thoroughly until its washing was no longer alkali. The residue was then introduced into an already dried crucible and ashed at 550°C (Oluwajoba *et al.*, 2013).

#### **Fat Content Determination**

Two grams of the sample was loosely wrapped with a filter paper and put into the thimble which was fitted to a clean round bottom flask, which has been cleaned, dried and weighed. The flask contained 120 ml of petroleum ether. The sample was heated with a heating mantle and allowed to reflux for 5 h. The heating was then stopped and the thimbles with the spent samples kept and later weighed. The difference in weight was received as mass of fat and is expressed in percentage of the sample (Nwosu *et al.*, 2011).

#### **Crude Protein Determination**

The micro kjeldahl method described by A.O.A.C (1990) was used. Two grams of each of the samples was mixed with 10 ml of concentrated H<sub>2</sub>SO<sub>4</sub> in a heating tube. One tablet of selenium catalyst was added to the tube and the mixture was heated inside a fume cupboard. The digest was transferred into distilled water. Ten millimeter portion of the digest was mixed with equal volume of 45% NaOH solution and poured into a kjeldahl distillation apparatus. The mixture was distilled and the distillate collected into 4% boric acid solution containing drops of methyl red indicator. A total of 50 ml distillate was collected and titrated as well. The Nitrogen content was calculated and multiplied with 6.25 to obtain the crude protein content.

#### **Determination of Carbohydrate**

This was determined as the difference obtained after subtracting total organic nitrogen (protein), Lipid, Ash, Moisture and Fibre from the total dry matter (AOAC., 2005).

### **Determination of Anti-nutritional Content of Natural Fermented *Obiolor***

#### **Determination of Phytate**

The method described by AOAC (2002) was used for phytate determination. Sample (2 g) was weighed into 250 ml conical flask. 100 ml of 2% concentrated hydrochloric acid was used to soak each of the samples in conical flasks for 3 h and then filtered through a double layer filter paper. 50 ml of each of the sample filtrates was placed in 250 ml beakers and 100 ml of distilled water was added to each of the samples to improve proper acidity. 10 ml of 0.3% Ammonium Thiocyanate solution was added to each sample solution as indicator and titrated with standard iron (III) chloride solution which contained 0.00195 g iron per ml. The end point was signified by brownish-yellow coloration that persisted for 5 mins (Adelekan *et al.*, 2013).

#### **Determination of Trypsin Inhibitor**

Samples (0.2 g) were weighed into screw cap centrifuge tubes. 10 ml of 0.1 M phosphate buffers were added into each and contents were shaken at room temperature for 1 h on stuart orbital shaker. The suspension obtained was

thereafter centrifuged at 500 rpm for 5 mins and filtered through Whatman 125 mm filter paper. The volumes of each were adjusted with 2 ml phosphate buffer and the test tubes were placed in water bath maintained at 37°C. 6 mls of 5% Trichloroacetic acid (TCA) solution was added to an empty tube to serve as the blank. 2 mls of casein solution was added to each of the tubes initially placed in the water bath before incubation for 20 mins. 6 mls of TCA solution was added into the sample tubes 20 minutes after (so as to stop the reaction) and shaken. The reaction was allowed to proceed for 1 hour at room temperature and the mixture was filtered through Whatman 125 mm filter paper. The absorbance of sample filtrates and trypsin standard solutions were then read at 280 nm (Adelekan *et al.*, 2013)

#### Statistical Analysis

Analysis of variance (ANOVA) was used to determine statistically significant differences among treatments using SPSS statistical soft-ware, version 17.

## Results and Discussion

The pH of *Obiolor* samples ranged from 4.80-3.80 for wet-milled sample (Figure 1) while pH for flour-milled prepared *Obiolor* ranged from 4.70-4.00 from 6 to 24 h of fermentation period respectively. There was a reduction in pH during the natural fermentation of sorghum and millet for *Obiolor* production. The TTA of the *Obiolor* beverage during fermentation is illustrated in Figure 2. There was a steady increase in the TTA in the range of 0.12% to 0.17% during 24 h fermentation period for wet-milled prepared *Obiolor* while the flour-milled *Obiolor* had its values in the range of 0.14%-0.22% from 6-24 h fermentation time. This trend has been reported in other African fermented drinks such as the fermentation of millet-acha based *kunun-zaki* beverage (Ayo, 2004) and fermentation of maize grains for *Ogi* production (Oyediji *et al.*, 2013) amongst several others. The reduction in pH and increase in TTA could be attributed to the actions of lactic acid bacteria (Ojokoh *et al.*, 2015). It is a beneficial factor for the fermenting organisms in which only acid tolerant microorganisms will thrive at the end of fermentation (Muyanja *et al.*, 2010).

Regardless of the significant rise in acidity, total soluble solids remained stable. Total Soluble Solids of the *Obiolor* samples is shown in Table 1. TSS was 9.00 °brix throughout the period of natural fermentation for *Obiolor* production. TSS was also reported to have total soluble solids of 8 to 9 °brix in *Tchapalo*, an Ivorian sorghum beverage (Djè *et al.*, 2008). The reduced amount of TSS contents during *Obiolor* fermentation could be attributed to the quick consumption of available solids carried out by yeast. It could also be as a result of action of fermentative microbes on the carbohydrate of the filtrate as the solid has been partially gelatinized (Danbaba *et al.*, 2014).

The specific gravity of the beverage is shown in Table 2. Wet-milled prepared *Obiolor* had a specific gravity of 1.009 while flour-milled prepared sample had a specific gravity of 1.008. These values remain constant throughout the period of fermentation. This result is related to what was obtained by Odibo *et al.* (2007), who found the specific gravity of beverage from two sorghum varieties to be 1.008 and 1.003 respectively.

#### Nutritional Analysis

The proximate composition of *Obiolor* is presented in Table 3. Proximate analysis is a vital tool in the assessment of nutritional status of food and food products (Ajiboye *et al.*, 2014). Moisture content of naturally fermented *Obiolor* had high values in the range of 74.41±0.15 to 71.79±0.00% (P<0.05) for both wet and flour-milled *Obiolor* (Table 5). *Kunu-zaki*, another cereal non-alcoholic beverage was found to be in the range of 82.0±0.15% to 90.70±0.15%. The high levels of moisture content are expected because it is a liquid based beverage (Ajiboye *et al.*, 2014). The ash content of wet-milled prepared *Obiolor* is 2.86±0.00% while the ash content for flour-milled prepared *Obiolor* is 1.42±0.00%. This value is in close range with 2.40±0.03 obtained by Ajiboye *et al.* (2014), however, values are lower than what was obtained by Ekanem *et al.* (2018) who worked on *Kunu* drink, a non alcoholic cereal beverage. Fat content in this study was also not significantly different (P>0.05) in *Obiolor* prepared from wet milled process (5.20±0.20%) and flour- milled process (5.38±0.12%). These values are higher than 0.39±0.01% obtained by Ajiboye *et al.* (2014) but lower than 5.5% obtained by Ekanem *et al.* (2018). Studies have reported that germination reduces fat content (Jan *et al.*, 2017) due to the break down and consumption of fats as an energy source for biochemical reactions during germination (Jan *et al.*, 2017). The protein content for wet and flour-milled prepared beverages were 4.40±0.01% and 3.89±0.04% respectively. Essien *et al.* (2011) reported that loss of protein during processing of the drinks may be responsible for the low content observed. Ofudje *et al.* (2016) reported that the protein content of *Kunu-zaki*, a non-alcoholic local beverage was revealed to be in the range of 2.18±0.02 to 8.40±0.23%. The consequence of fermentation on protein has yielded inconsistent results possibly due to diverse experimental designs, study durations and variation in the initial protein content of foods. A number of studies had reported increase (Doudu *et al.*, 2003; Pranoto *et al.*, 2013) while others decrease (Osman, 2011; Pranoto *et al.*, 2013) in protein value upon fermentation. It seems that most of these effect caused relative changes due to loss of dry matter as a result of microorganisms hydrolysing and metabolizing carbohydrates as source of energy (Smith *et al.*, 2018). Carbohydrate content of *Obiolor* observed in this study was 11.10±0.22 and 13.52±0.60 (P<0.05) for *Obiolor* prepared from both wet and flour-milled process respectively. The

carbohydrate reduction during fermentation shows the beverage as an ideal substrate for microorganisms. The main carbohydrate in cereal beverages is starch which provides the most calories in developing countries (Chaves-Lopez *et al.*, 2014). The quantity of available carbohydrate in the beverage could serve as a source of energy from adenosine triphosphate (Ajiboye *et al.*, 2014). There was no significant difference in the crude fibre for both wet and flour-milled *Obiolor* which was  $1.75 \pm 0.25\%$  and  $3.50 \pm 0.50\%$  respectively. These values are higher than  $0.30 \pm 0.01$  obtained from sorghum-millet beverage (Ajiboye *et al.* 2014) and 1.001 obtained from *kunu drink* (Ekanem *et al.* 2018).

#### **Antinutritional Analysis**

The phytic acid and trypsin inhibitor of *Obiolor* were recorded after fermentation as showed on Table 6. The phytic acid concentration of wet-milled prepared *Obiolor* was 1.20 mg/100 g while *Obiolor* from flour-milled process had a phytic acid concentration of 1.00 mg/100 g. These values have higher phytic acid concentration than malted soy *Kunu-zaki* (Adelekan *et al.*, 2013) which had a lower value to be 1.00 mg/100 g in flour-milled prepared beverage. However, a different trend was observed where the concentration of wet-milled prepared beverage had trypsin inhibitor concentration of 3.426 mg/100 g which was lower than the value of 3.488 mg/100 g obtained for flour-milled prepared beverage. Germination has been reported to diminish the concentration of antinutritional factors in grains. Fermentation also enhances bioavailability of calcium, phosphorus and iron likely due to degradation of oxalates and phytates that combine with minerals thereby reducing their bioavailability hence, making them free and more accessible (Sripriya *et al.*, 1997).

#### **Conclusion**

In this study, *Obiolor* was produced from sorghum and millet grains using two different methods: wet-milled and flour-milled prepared *Obiolor*. The results were slightly different from each other except for total soluble solids which had the same values for both wet and flour-milled beverages. However, proximate analysis revealed that flour-milled *Obiolor* had more protein, carbohydrate, crude fibre and fat content, indicating that it is of more nutritive value than wet-milled produced *Obiolor* which is commonly practised at household level. The use of sorghum-millet flour also allows for preservation of the grains for production of *Obiolor*. Further studies need to be conducted to investigate the effect of different steeping and germination conditions for *Obiolor* production.

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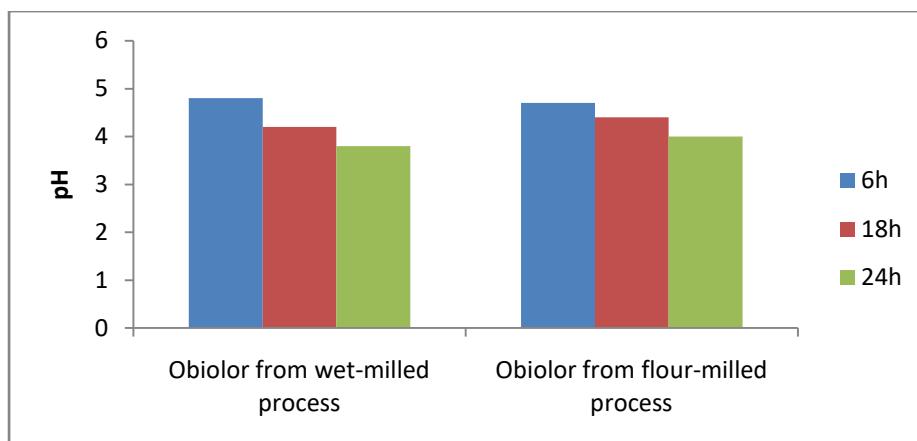


Figure 1: pH of Sorghum and Millet during Natural Fermentation of *Obiolor*

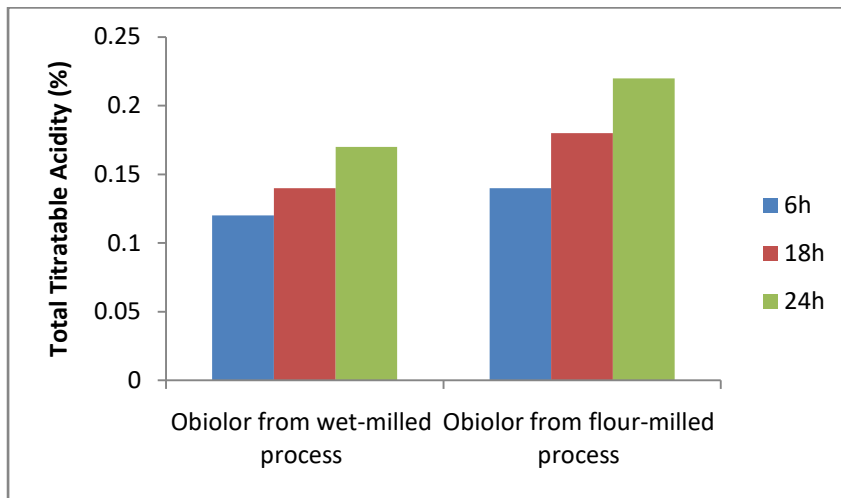


Figure 2: Total titratable acidity of Sorghum and Millet during Natural Fermentation of *Obiolor*

Table 1: Total Soluble Solids of Naturally Fermented *Obiolor*

Sorghum/millet Beverage	°Brix
<i>Obiolor</i> from wet-milled process	9.00
<i>Obiolor</i> from dried-milled process	9.00

Values are mean  $\pm$  SD, where n=3. Mean with different superscript across the same row are significantly different (p<0.05).

Table 2: Specific Gravity of Naturally Fermented *Obiolor*

Sorghum/millet Beverage	Specific Gravity
<i>Obiolor</i> from wet-milled process	1.009
<i>Obiolor</i> from flour-milled process	1.008

Values are mean  $\pm$  SD, where n=3. Mean with different superscript across the same row are significantly different (p<0.05).



Table 3: Nutritional components of Naturally Fermented *Obiolor*

Sorghum and Millet Beverage	%Moisture	%Ash	%Fat	% Crude Fibre	% Crude Protein	%CHO
<i>Obiolor</i> from wet-milled process	74.41±0.15 <sup>a</sup>	2.86±0.00	5.20±0.02 <sup>a</sup>	1.75±0.25 <sup>a</sup>	3.89±0.04 <sup>a</sup>	11.10±0.22 <sup>a</sup>
<i>Obiolor</i> from flour-milled Process	71.79±0.00 <sup>b</sup>	1.42±0.00	5.38±0.12 <sup>a</sup>	3.50±0.50 <sup>a</sup>	4.40±0.01 <sup>b</sup>	13.52±0.60 <sup>a</sup>

Values are mean ± SD, where n=3. Mean with different superscript across the same row are significantly different (p<0.05).

Key: CHO – Carbohydrate

Table 4: Antinutritional Component of Naturally Fermented *Obiolor*

Sorghum/millet Beverage	Phytic acid (mg/100 g)	Trypsin Inhibitor (mg/100 g)
<i>Obiolor</i> from wet-milled process	1.20	3.426
<i>Obiolor</i> from flour-milled process	1.00	3.488

Values are mean ± SD, where n=3. Mean with different superscript across the same row are significantly different (p<0.05).